

The Influence of Different Ammonium Cations on the Optical Properties of Tetrakis Gd^{III} and Eu^{III} Complexes

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Table S1. Infrared frequencies (wavenumber) for the tta ligand and the Q⁺[Eu(tta)₄]⁻ complexes

Compound	$\nu(\text{C-H}) / \text{cm}^{-1}$	$\nu_a(\text{C=O}) / \text{cm}^{-1}$	$\nu_s(\text{C=O}) / \text{cm}^{-1}$	$\nu(\text{C=C}) / \text{cm}^{-1}$	$\nu(\text{C-O}) / \text{cm}^{-1}$	$\nu_s(\text{C-F}) / \text{cm}^{-1}$
tta	3720-2289	1691	1632	1541	1353	1295
(N(C ₂ H ₅) ₄) ⁺ [Eu(tta) ₄] ⁻	2989	–	1605	1535 1507 1481	1356	1183 1133
(N(C ₄ H ₉) ₄) ⁺ [Eu(tta) ₄] ⁻	2967 2878	–	1606	1535 1504 1480	1356	1183 1133
(N(C ₁₂ H ₂₅) ₂ (CH ₃) ₂) ⁺ [Eu(tta) ₄] ⁻	2956	–	1616	1540	1356	1308

tta: 2-thenoyltrifluoroacetone; bmdm: 1-(4-*tert*-butylphenyl)-3-(4-methoxyphenyl)propane-1,3-dione.

Table S2. Infrared frequencies for the bmdm ligand and the Q⁺[Eu(bmdm)₄]⁻ complexes

Compound	$\nu(\text{C-H}) / \text{cm}^{-1}$	$\nu_a(\text{C=O}) / \text{cm}^{-1}$	$\nu_s(\text{C=O}) / \text{cm}^{-1}$	$\nu(\text{C=C}) / \text{cm}^{-1}$	$\nu(\text{C-O}) / \text{cm}^{-1}$	$\nu_a(\text{C-O-C}) / \text{cm}^{-1}$
bmdm	3178 2950	1612	1586	1500	1353	1259 1182
(N(C ₂ H ₅) ₄) ⁺ [Eu(bmdm) ₄] ⁻	3238 2960	1594	1569	1527 1488 1434 1431 1461	1356	1254 1171
(N(C ₄ H ₉) ₄) ⁺ [Eu(bmdm) ₄] ⁻	3238 2960	1594	1569	1439 1458	1356	1254 1171
(N(C ₁₂ H ₂₅) ₂ (CH ₃) ₂) ⁺ [Eu(bmdm) ₄] ⁻	3238 2960	1594	1569	1439 1458	1356	1254 1171

tta: 2-thenoyltrifluoroacetone; bmdm: 1-(4-*tert*-butylphenyl)-3-(4-methoxyphenyl)propane-1,3-dione.

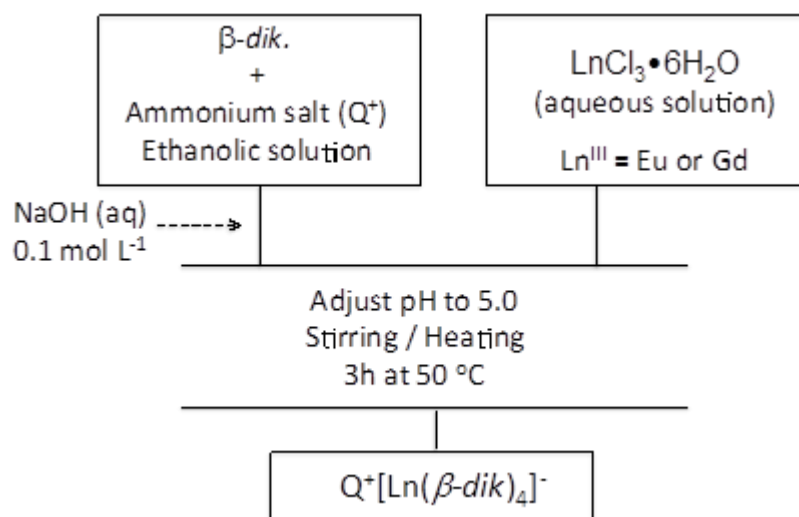


Figure S1. Synthetic route to the obtention of $\text{Q}^+[\text{Ln}(\beta\text{-dik})_4]^-$.

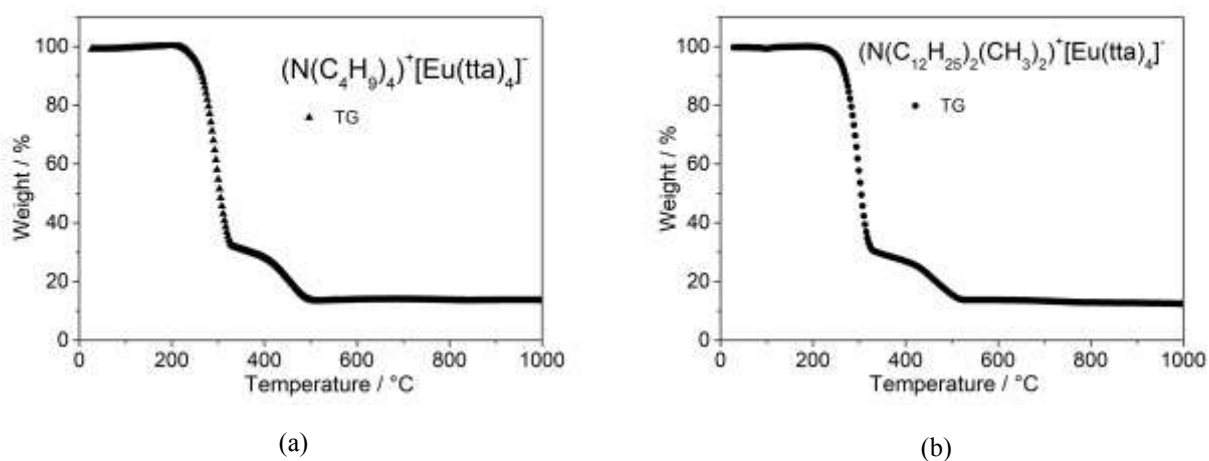


Figure S2. TG curves of (a) $(\text{N}(\text{C}_4\text{H}_9)_4)^+[\text{Eu}(\text{tta})_4]^-$ and (b) $(\text{N}(\text{C}_{12}\text{H}_{25})_2(\text{CH}_3)_2)^+[\text{Eu}(\text{tta})_4]^-$ complexes.

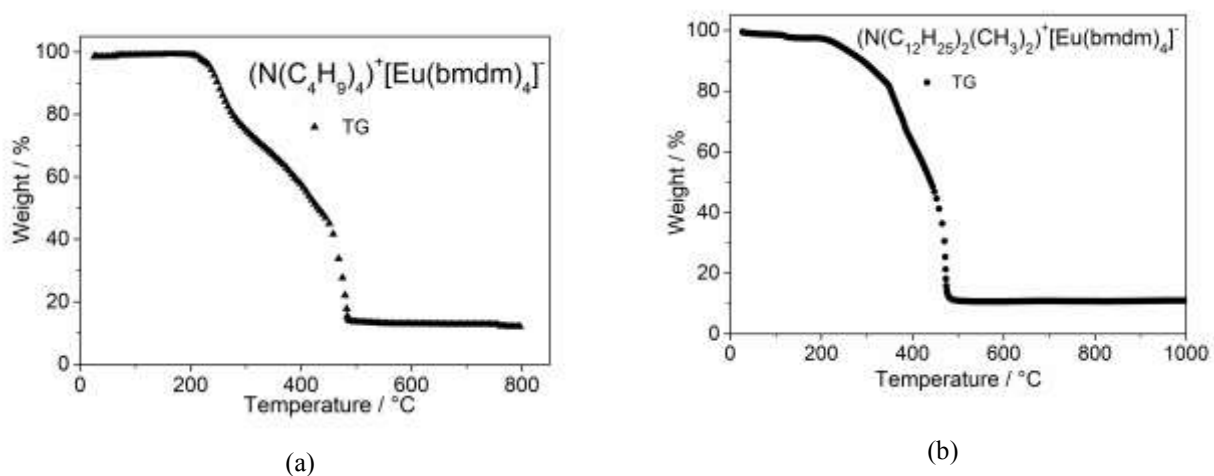
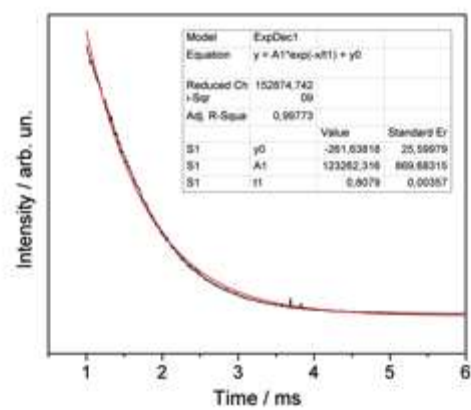
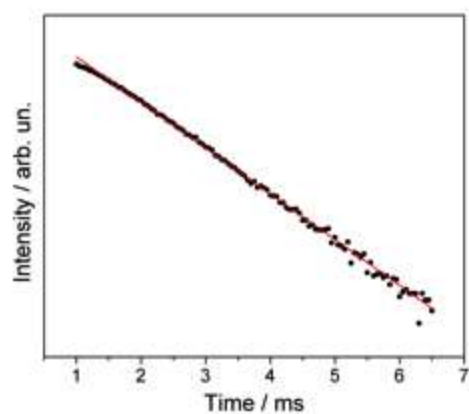


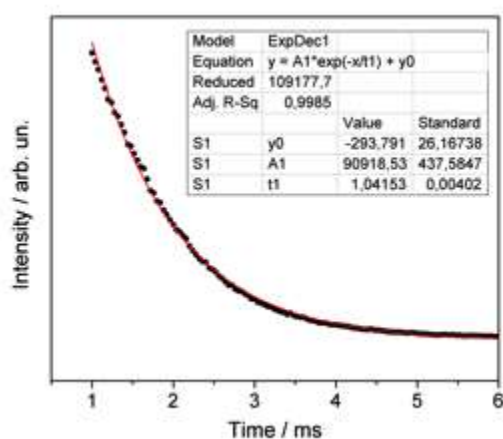
Figure S3. TG curves of (a) $(\text{N}(\text{C}_4\text{H}_9)_4)^+[\text{Eu}(\text{bmdm})_4]^-$ and (b) $(\text{N}(\text{C}_{12}\text{H}_{25})_2(\text{CH}_3)_2)^+[\text{Eu}(\text{bmdm})_4]^-$ complexes.



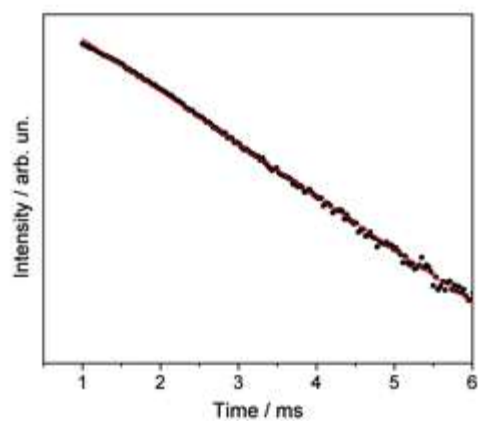
(a)



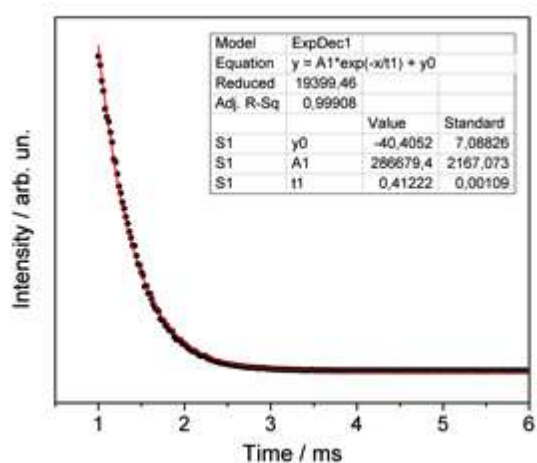
(b)



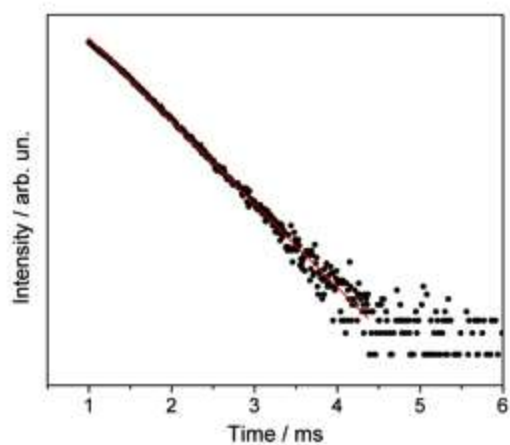
(c)



(d)

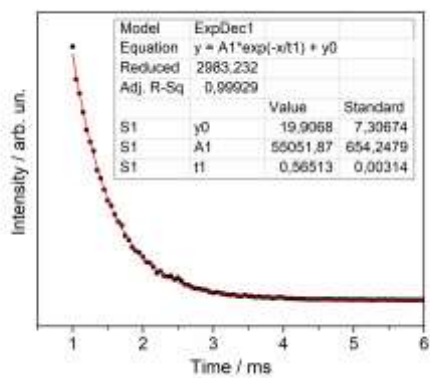


(e)

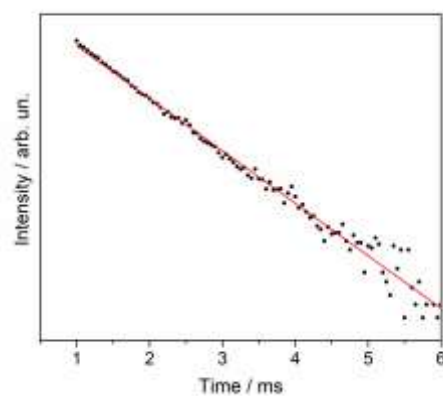


(f)

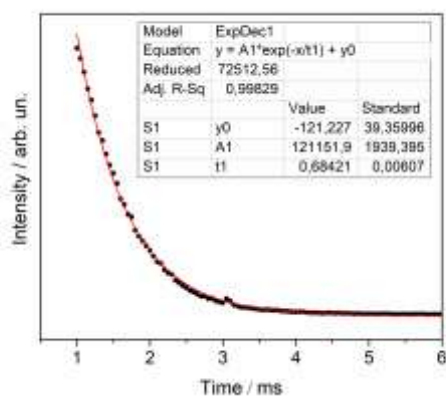
Figure S4. Emission decay curves and linearization for $Q^+[\text{Eu}(\text{tta})_4]^-$. (a, b) $Q = (\text{N}(\text{C}_2\text{H}_5)_4)^+$; (c, d) $Q = (\text{N}(\text{C}_4\text{H}_9)_4)^+$; (e, f) $Q = (\text{N}(\text{C}_{12}\text{H}_{25})_2(\text{CH}_3)_2)^+$.



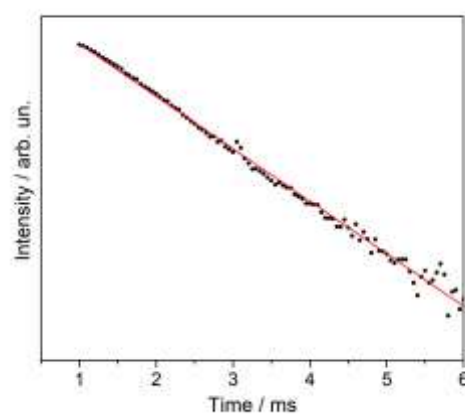
(a)



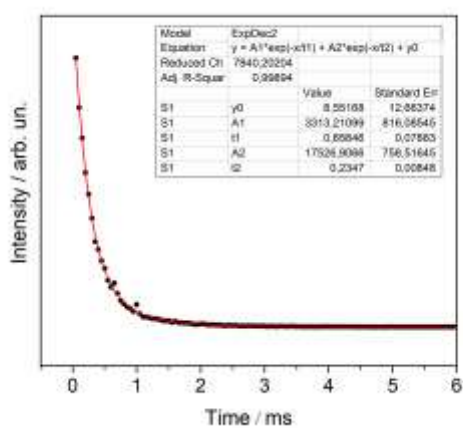
(b)



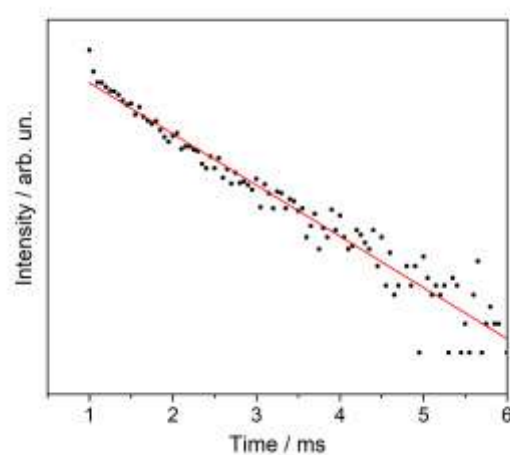
(c)



(d)



(e)



(f)

Figure S5. Emission decay curves and linearization for $Q^+[\text{Eu}(\text{bmdm})_4]^-$. (a, b) $Q = (\text{N}(\text{C}_2\text{H}_5)_4)^+$; (c, d) $Q = (\text{N}(\text{C}_4\text{H}_9)_4)^+$; (e, f) $Q = (\text{N}(\text{C}_{12}\text{H}_{25})_2(\text{CH}_3)_2)^+$.

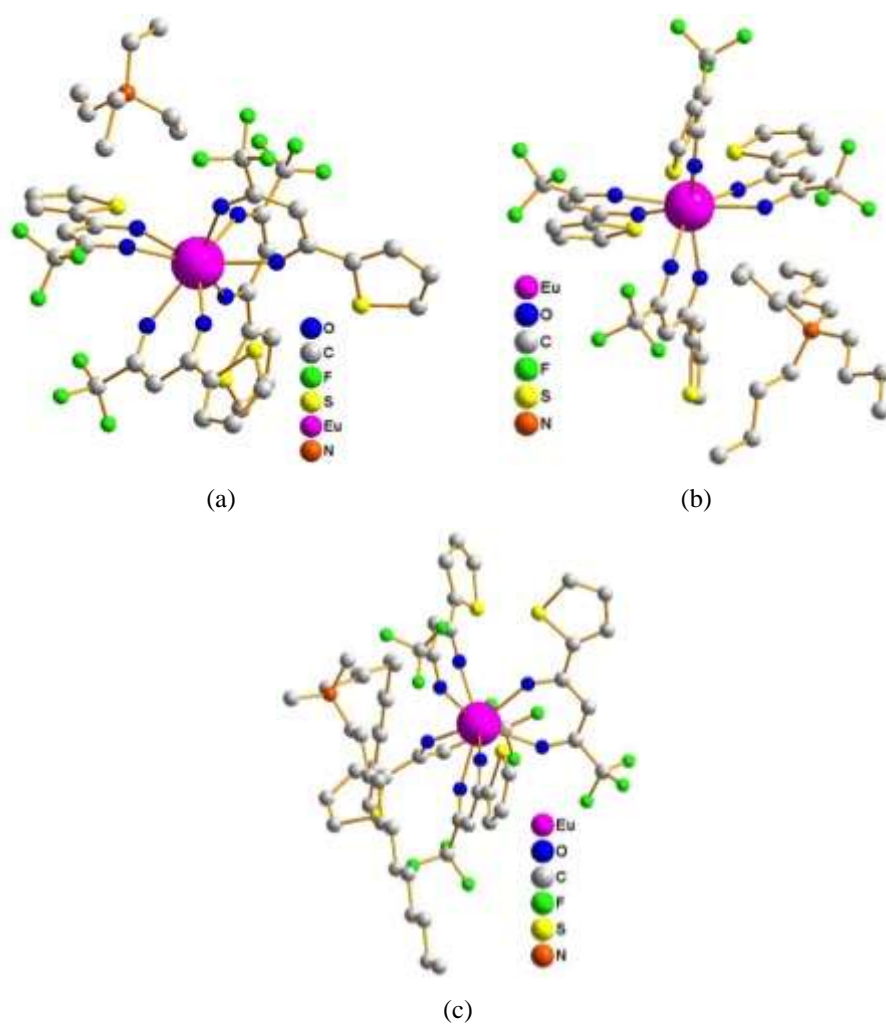


Figure S6. Ground state geometry of the (a) $(\text{N}(\text{C}_2\text{H}_5)_4)^+[\text{Eu}(\text{tta})_4]^-$; (b) $(\text{N}(\text{C}_4\text{H}_9)_4)^+[\text{Eu}(\text{tta})_4]^-$ and (c) $(\text{N}(\text{C}_{12}\text{H}_{25})_2(\text{CH}_3)_2)^+[\text{Eu}(\text{tta})_4]^-$ complexes obtained by the Sparkle/PM6.

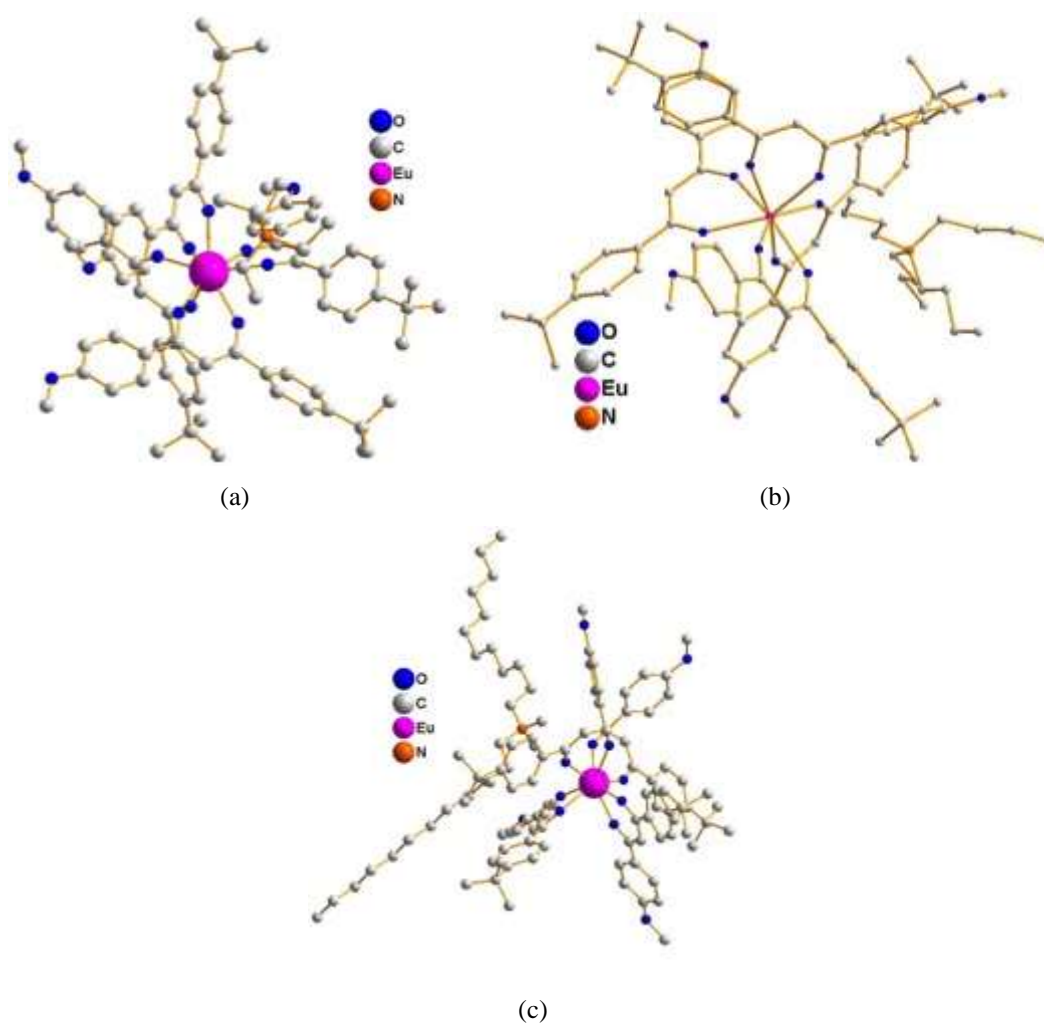


Figure S7. Ground state geometry of the (a) $(\text{N}(\text{C}_2\text{H}_5)_4)^+[\text{Eu}(\text{bmdm})_4]^-$; (b) $(\text{N}(\text{C}_4\text{H}_9)_4)^+[\text{Eu}(\text{bmdm})_4]^-$ and (c) $(\text{N}(\text{C}_{12}\text{H}_{25})_2(\text{CH}_3)_2)^+[\text{Eu}(\text{bmdm})_4]^-$ complexes obtained by the Sparkle/PM6.