

Oxidation of Mono-Phenols to *para*-Benzoquinones: a Comparative Study

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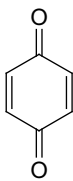
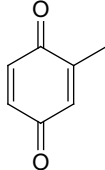
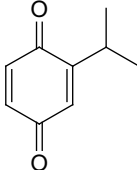
General procedure for the oxidations with H₂O₂ and Br₂

A solution of the phenol (1 mmol) and Br₂ (0.5 mmol) was added to a solution of H₂O₂ (30%, 0.2 mL, 4 mmol,) and concentrated H₂SO₄ (0.1 mL) in methanol (3 mL). The resulting mixture was refluxed for 2 hours, the total volume then reduced to eliminate excess of methanol, extracted with ether, and dried over anhydrous MgSO₄. The solvent was evaporated, and the *para*-benzoquinone product was isolated by flash chromatography on silica gel using as eluent a mixture of 9:1 hexane: ethyl acetate.

General procedure for the oxidations with IBX

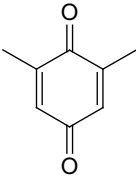
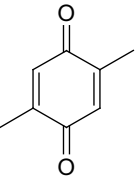
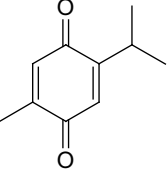
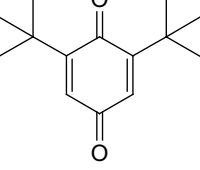
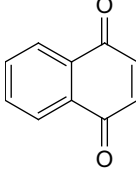
The phenol (0.3 mol) was dissolved in 5.7 mL of anhydrous CH₂Cl₂ or anhydrous DMF, IBX (0.3 mol) was added, and the mixture stirred for 24 hours. The reaction mixture was diluted with EtOAc (20 mL), washed with water (10 mL), and this aqueous phase extracted with EtOAc (2 × 10 mL). The organic phase was washed with brine, dried over anhydrous MgSO₄ and concentrated.

Table S1. NMR spectral data and melting points of the *para*-benzoquinone products

Product	¹ H NMR (CDCl ₃ /TMS) δ, J(Hz)	¹³ C NMR (CDCl ₃ /TMS) δ	mp / °C	Lit. mp / °C
 12	6.79 (4H, s)	136.6, 187.2	110-114	113-115 ¹⁶
 13	1.90 (CH ₃ ,s), 6.62 (CH, s), 6.75 (CH, d), 6.77 (CH, d)	15.8, 133.3, 136.4, 136.5, 145.9, 187.5, 187.7	65-68	67-70 ²⁶
 14	1.14 (2CH ₃ , d, J 6.8 Hz), 3.05 (CH, d, J 6.8; 1.1 Hz), 6.55 (CH, m, J 1.1 Hz), 6.73 (CH, d, J 2.5 Hz), 6.74 (CH, d, J 2.5 Hz)	16.5, 26.2, 130.3, 135.9, 137.0, 154.9, 187.1, 188.1	57-60	54-58 ¹⁶

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Table S1. (cont.)

Product	$^1\text{H NMR}$ (CDCl_3/TMS) δ , $J(\text{Hz})$	$^{13}\text{C NMR}$ (CDCl_3/TMS) δ	mp / $^\circ\text{C}$	Lit. mp / $^\circ\text{C}$
 15	2.10 (6H, m), 6.56 (2H, m)	15.9, 133.3, 145.7, 187.1	69-72	71-73 ²⁷
 16	2.04 (6H, d, J 1.6 Hz), 6.60 (2H, q, J 1.6 Hz)	15.6, 133.3, 145.7, 188.7	121-123	124-125 ²²
 17	1.13 (6H, d, J 6.8 Hz), 2.04 (3H, d, J 1.6 Hz), 3.03 (1H, d hept, J 1.2 e 6.8 Hz), 6.52 (1H, d, J 1.2 Hz), 6.59 (1H, q, J 1.6 Hz).	15.3, 21.4, 26.5, 130.3, 133.8, 145.1, 154.9, 187.4, 188.5	47-48	45-47 ²⁰
 18	1.28 (CH_3 , s), 6.51 (CH, s)	29.3, 35.5, 130.1, 157.9, 188.7, 189.0	64-66	65-68 ³⁴
 19	8.09 (2CH, aa'bb'), 7.77 (2CH, aa'bb'), 6.98 (2CH, s)	126.4, 133.9, 134.1, 138.7, 185.3	118-120	120-122 ²⁷