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Two New Copper(II) Complexes of 1-Butyl-1*H*-1,2,4-triazole: Synthesis, Characterization and Electrocatalytic Activity

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*Electrocatalytic activity of the **1**-CPE modified electrode*

The electrocatalytic activities of **1**-CPE toward the reduction of hydrogen peroxide and trichloroacetic acid are presented in Figures S1 and S2. The detection limit, linear calibration range and the sensitivity are respectively 0.01 μmol L^{-1}, 0.02 to 0.20 μmol L^{-1} and 84.45 μA μmol L^{-1} for hydrogen peroxide detection, and 0.01 μmol L^{-1}, 0.02 to 0.15 μmol L^{-1} and 84.73 μA μmol L^{-1} for TCA detection. These results indicate that **1**-CPE and **2**-CPE show similar cyclic voltammetry behaviors and electrocatalytic activities.

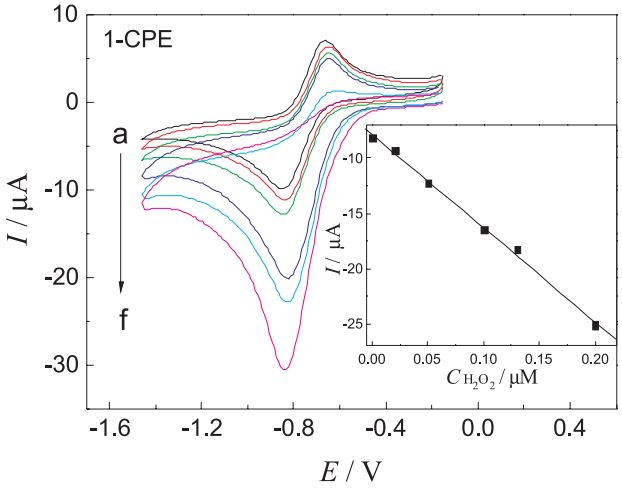

Figure S1. Cyclic voltammograms for **1**-CPE in 0.1 mol L^{-1} B-R buffer solution, pH 6.1, containing (a-f) 0, 0.02, 0.05, 0.10, 0.13 and 0.20 μmol L^{-1} H_2O_2. Scan rate: 0.05 V s^{-1}. Inset: cathodic peak current *vs*. H_2O_2 concentration.

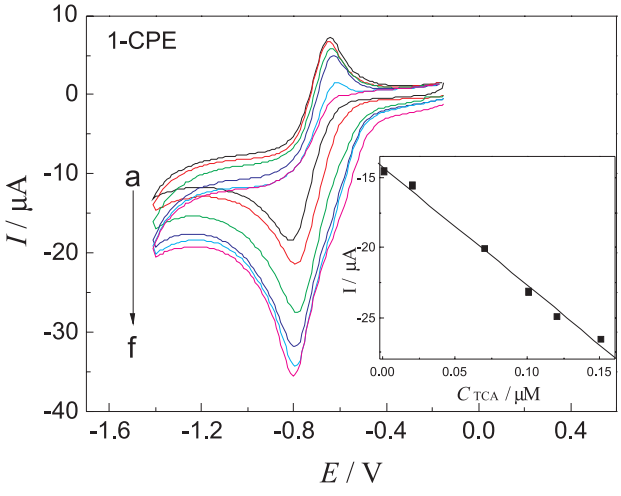

Figure S2. Cyclic voltammograms for **1**-CPE in 0.1 mol L^{-1} B-R buffer solution, pH 6.1, containing (a-f) 0, 0.02, 0.07, 0.10, 0.12 and 0.15 μmol L^{-1} TCA. Scan rate: 0.05 V s^{-1}. Inset: cathodic peak current *vs*. TCA concentration.

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