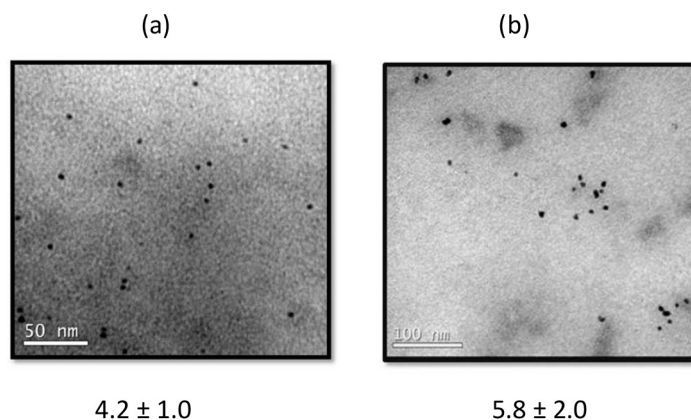


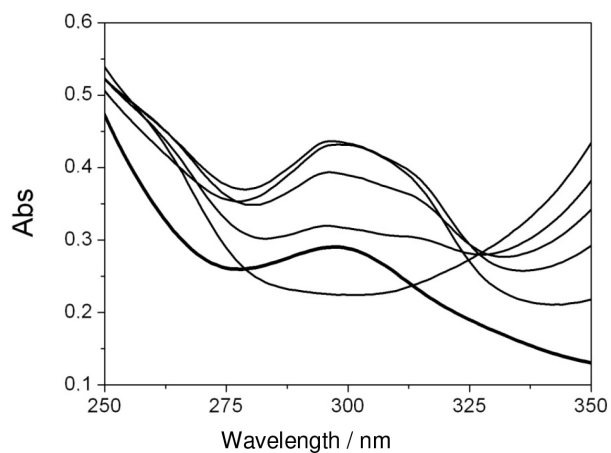
## Zwitterionic Surfactant Stabilized Palladium Nanoparticles as Catalysts in Aromatic Nitro Compound Reductions

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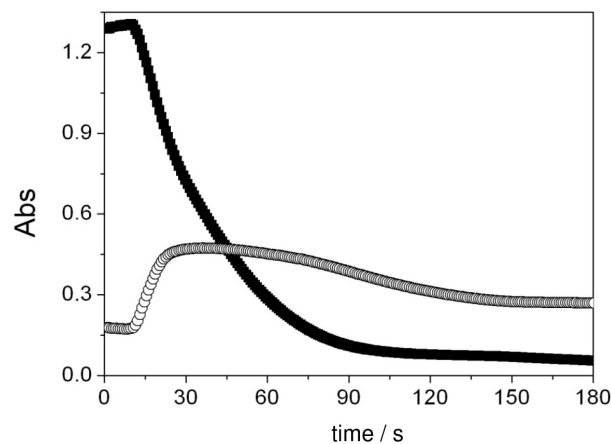
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**Figure S1.** TEM microscopy and the average diameter (of approximately 300 particles) of palladium nanoparticles synthesized in: (a)  $w_0$  4.6 (value taken from reference 1) and (b)  $w_0$  14.7 (value taken from reference 2).

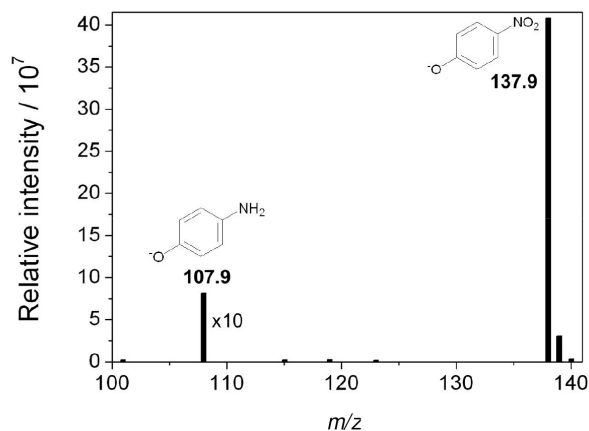


**Figure S2.** Variation of UV-Vis absorption spectra of reduction of 4-nitrophenol ( $0.066 \text{ mmol L}^{-1}$ ) with  $\text{NaBH}_4$  ( $9.71 \text{ mmol L}^{-1}$ ) catalyzed by palladium nanoparticles (Pd NPs  $1-0.0167 \text{ mmol L}^{-1}$ ) ( $[\text{NaCl}] = 0.033 \text{ mmol L}^{-1}$ ). The thicker line refers to the end of the reaction.

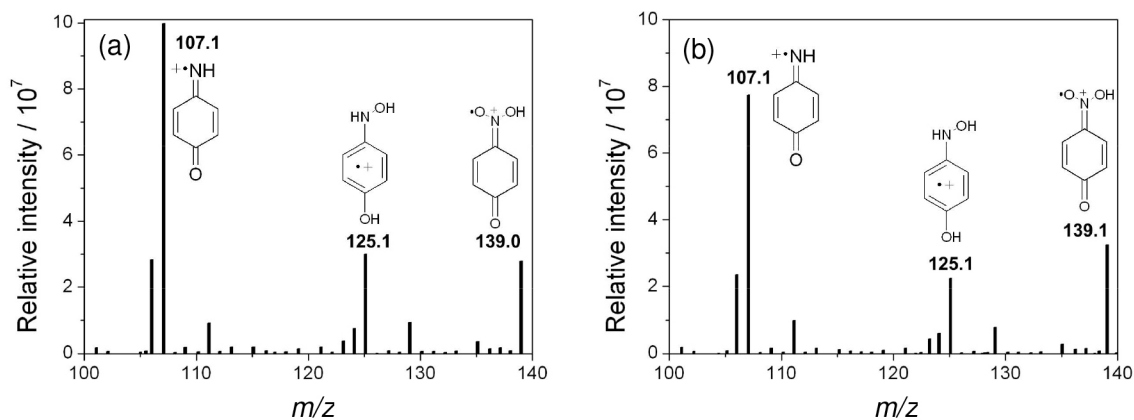


**Figure S3.** Time-dependence of the absorption of 4-nitrophenolate ions at (■) 400 nm and (○) 296 nm ( $[\text{4-nitrophenol}] = 0.066 \text{ mmol L}^{-1}$ ;  $[\text{NaBH}_4] = 9.71 \text{ mmol L}^{-1}$ ;  $[\text{Pd NPs 1}] = 0.0167 \text{ mmol L}^{-1}$ ,  $[\text{NaCl}] = 0.033 \text{ mmol L}^{-1}$ ). The reactions were carried out at  $25^\circ\text{C}$ .

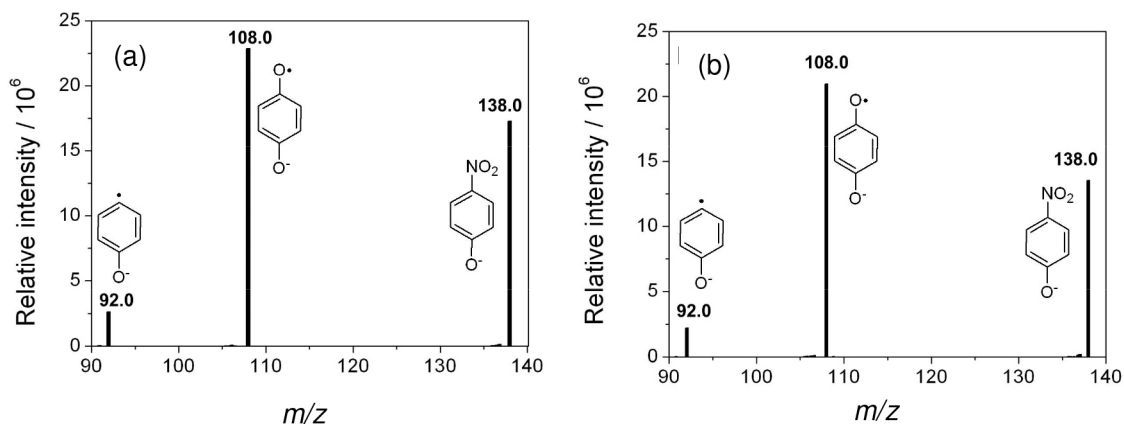
\*e-mail: franciane\_dutra@hotmail.com, faruk.nome@ufsc.br



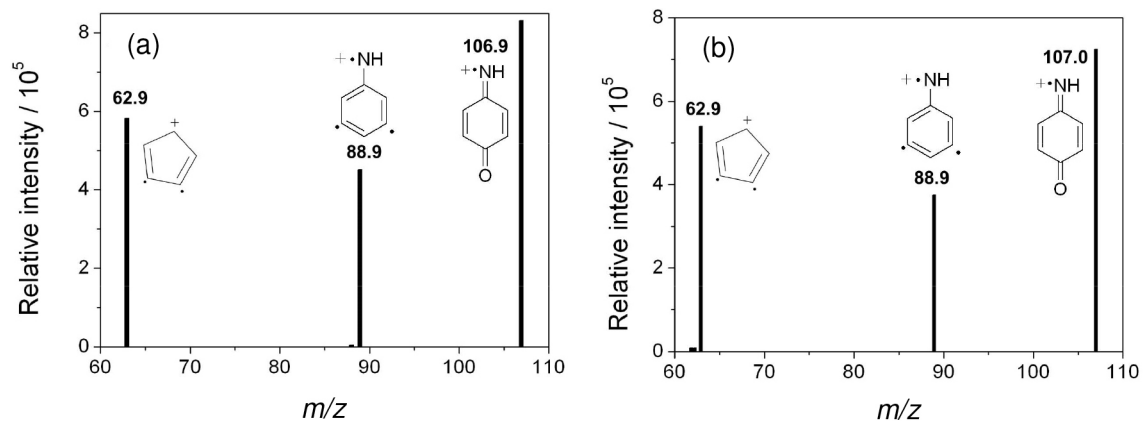
**Figure S4.** ESI-MS spectrum in negative mode of an aliquot taken after approximately 5 min of the mixing of reactants in the presence of Pd NPs. The initial conditions were  $[\text{Nip}] = 0.093 \text{ mmol L}^{-1}$ ;  $[\text{NaBH}_4] = 9.71 \text{ mmol L}^{-1}$ ;  $[\text{NaCl}] = 0.033 \text{ mmol L}^{-1}$ ;  $[\text{Pd}] = 0.0019 \text{ mmol L}^{-1}$ . After addition of  $\text{NaBH}_4$  solution, the solution was diluted ten times and injected into the mass spectrometer.



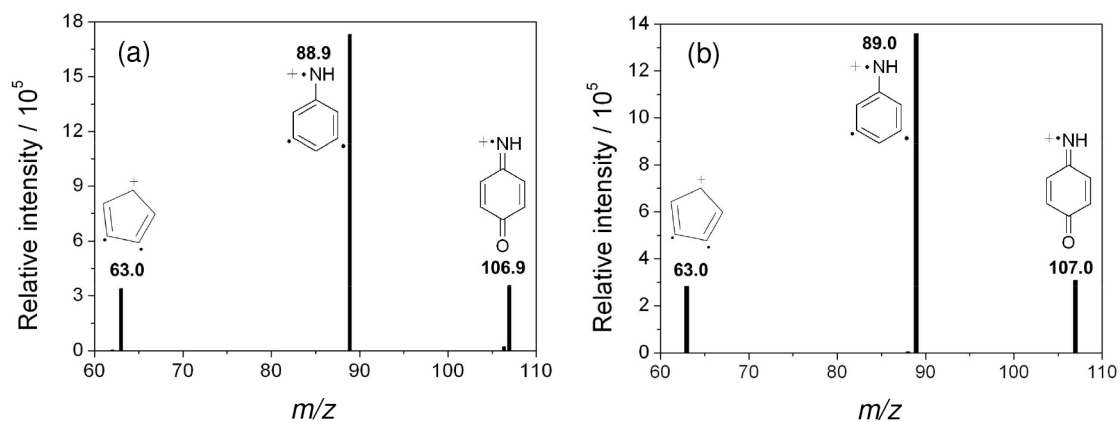
**Figure S5.** ESI-MS spectrum in positive mode of an aliquot taken after approximately 5 min of the mixing of reactants in the presence of Pd NPs. The initial conditions were  $[\text{Nip}] = 0.093 \text{ mmol L}^{-1}$ ;  $[\text{NaBH}_4] = 9.71 \text{ mmol L}^{-1}$ ;  $[\text{NaCl}] = 0.033 \text{ mmol L}^{-1}$ ;  $[\text{Pd}] = 0.0019 \text{ mmol L}^{-1}$ . After addition of  $\text{NaBH}_4$  solution, the solution was diluted ten times and injected into the mass spectrometer.



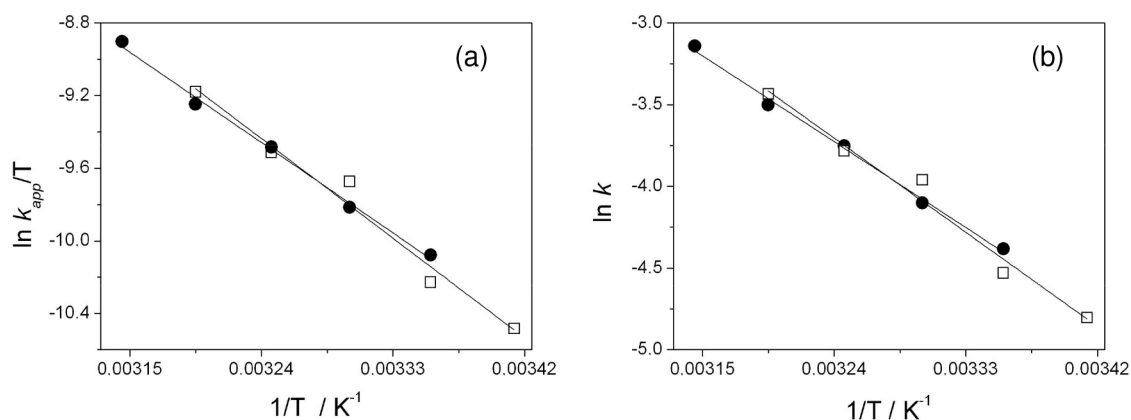
**Figure S6.** ESI-MS/MS of the intermediate anion of  $m/z$  138 in negative mode: (a) Pd NPs 2; (b) Pd NPs 2.



**Figure S7.** ESI-MS/MS of the intermediate anion of  $m/z$  125 in positive mode: (a) Pd NPs 2; (b) Pd NPs 2.



**Figure S8.** ESI-MS/MS of the intermediate anion of  $m/z$  107 in positive mode: (a) Pd NPs 2; (b) Pd NPs 2.



**Figure S9.** (a) Eyring and (b) Arrhenius plots for determination of activation parameters for reaction systems at  $[4\text{-nitrophenol}] = 0.093 \text{ mmol L}^{-1}$ ;  $[\text{BH}_4^-] = 9.71 \text{ mmol L}^{-1}$ ;  $[\text{NaCl}] = 0.033 \text{ mmol L}^{-1}$ ,  $[\text{Pd}] = 0.0167 \text{ mmol L}^{-1}$ . (●) Pd NPs 1; (□) Pd NPs 2.

## References

1. Fernandes, S. C.; de Souza, F. D.; de Souza, B. S.; Nome, F.; Vieira, I. C.; *Sens. Actuators, B* **2012**, *173*, 483.
2. Souza, F. D.; Souza, B. S.; Tondo, D. W.; Leopoldino, E. C.; Fiedler, H. D.; Nome, F.; *Langmuir* **2015**, *31*, 3587.