

## Germacranolides from *Eremanthus elaeagnus*

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Por extração das folhas de *Eremanthus elaeagnus* foram obtidos quatro germacranólidos, cujas estruturas foram estabelecidas através de espectrometria de RMN, bem como os flavonóides 3-metoxi-quercetina e luteolina.

Extraction of the leaves of *Eremanthus elaeagnus* furnished four germacranolides, whose structure was established by NMR spectrometry, as well as the flavones quercetin 3-methyl ether and luteolin.

**Key words:** *Eremanthus elaeagnus*; *Compositae*; *Vernonieae*; *germacranolides*.

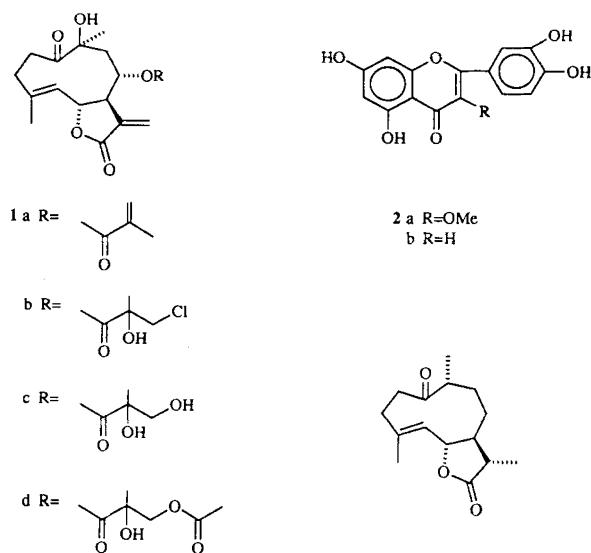
### Introduction

The Brazilian genus *Eremanthus* (Compositae, Vernonieae) as recently revised by MacLeish<sup>1,2</sup> contains 18 species, eight of which have been studied chemically<sup>3-21</sup>. Particular interest has centered on the schistosomicidal and cytotoxic constituents of the wood of *E. elaeagnus* Sch.Bip<sup>4,7,10</sup> and *E. erythropappa* (DC) MacLeish (syn. *Vanillosmopsis erythropappa* (DC) Sch.-Bip)<sup>5,6,9,19,21</sup>, which are sesquiterpene lactones of various types but other *Eremanthus* species also produce similar or identical lactones. In the present article we describe our work on the leaves of *Eremanthus elaeagnus* which have not been studied previously. Isolated were the germacranolides 1a-d as well as the common flavones 2a (luteolin) and 2b (quercetin 3-methyl ether).

The 500 MHz <sup>1</sup>H-NMR spectrum (Table 1) of 1a, C<sub>19</sub>H<sub>24</sub>O<sub>6</sub>, mp (180-182) °C, and selective decoupling permitted deduction of the entire carbon skeleton and stereochemistry. Thus irradiation at the frequencies of H-13a and H-13b whose appearance and chemical shifts (narrowly split doublets at δ 6.21 and 5.67) were characteristic of the γ-hydrogens of an α-methylene-γ-lactone (IR band at 1760 cm<sup>-1</sup>) simplified the multiplets of H7 at δ 3.27 which was in turn coupled to H-6, a *dd* at 4.60, and H-8, a *ddd* at δ 5.10. Irradiation at the frequency of H-8 simplified not only the signal of H-7 but also the mutually coupled signals of H-9a

and H-9b at δ 2.31 and 2.03 which were next to a quaternary center (C-10). Likewise, irradiation at the frequency of H-6 simplified not only the signal of H-7, but also the signal of H-5 at δ 5.17; the latter was allylically coupled to the protons of a vinylic methyl group at δ 1.91 and to the proton of a methylene group (H-3α) at δ 2.84, thus placing the vinylic methyl at C-4. Finally H-3α was not only coupled to its partner 3β, but also the protons of another methylene (H-2α-β) whose chemical shifts (δ 2.73 and δ 2.52 indicated that they were *alpha* to a ketone carbonyl. The presence of the latter was revealed by an IR band at 1698 cm<sup>-1</sup> and by a signal at δ 215.52 in the <sup>13</sup>C-NMR spectrum (Table 2). This necessitated placement of a second methyl group (three proton singlet at δ 1.22) and the tertiary hydroxyl (-OH absorption at δ 3.78), required by the molecular formula, on C-10 and was supported by the <sup>13</sup>C-NMR spectrum which in addition to doublets at δ 76.64 and 72.38 (C-6 and C-8) exhibited another signal in the C-O region, a singlet at δ 78.57 (C-10).

Chemical shifts and coupling constants of H-5, H-6, H-7 and H-8 were consonant with those of a 4*E*-germacren-6βH,12-olide containing an α-orientated ester function (IR bond at 1712 cm<sup>-1</sup>) on C-8; from the mass <sup>1</sup>H- and <sup>13</sup>C-NMR spectra (Tables 1 and 2) the ester was obviously a methacrylate.



As for the stereochemistry at C-10 it was originally assumed to be similar to that of other sesquiterpene lactones from *Eremanthus* and related species, i.e. -OH  $\alpha$  if C-10 is depicted in apical form as in formulas 1a-d. However, while the NOE difference spectrum (Table 3 exhibited the NOE's between H-3 $\alpha$  and H-5, H-5 and H-7, H-3 $\beta$  and H-15, H-6 and H-8 and H-6 and H-15 ordinarily displayed by an 8 $\alpha$ -acyloxygermacra-1(10),5,11(13)-6 $\beta$ H,12-olide or its equivalents in the usual crown conformation, it also exhibited NOE's between H-15, on the one hand, and H-2 $\beta$  and H-9 $\beta$  on the other and no NOE between the two methyl groups. This requires an arrangement in which the C-10 methyl group is  $\alpha$ -orientated and equatorial similar to its orientation in the crown conformation established for the closely related 3 by X-ray analysis<sup>22</sup>. In the conformation adopted by 1a and its congeners the dihedral angles between H-8 and H-9 $\alpha,\beta$  appear to be somewhat altered from those ordinarily prevailing, thus producing values of  $J_{8,9}$  (4 and 3Hz) different from those usually observed (10 and 1 Hz). The same structure, without stereochemistry at C-10, has been ascribed to a non crystalline incompletely characterized lactone from *Vernonia jonesii*<sup>23</sup>.

The three other lactones 1b-1d differed from 1a only in the nature of the C-8 ester side chain whose structures were obvious from the mass, <sup>1</sup>H- and <sup>13</sup>C-NMR spectra (Tables 1 and 2).

## Experimental

**Extraction of *Eremanthus elaeagnus*.** Leaves of *E. elaeagnus* Sch.Bip. (3.0 kg) were collected in Diamantina, Minas Gerais State, Brasil, in June 1985. The pulverized material was extracted with EtOAc to give 140 g of crude extract which was dissolved in MeOH-H<sub>2</sub>O (9:1). After extraction with hexane the aqueous methanol solution was concentrated at reduced pressure and extracted with EtOAc yielding 92.5 g of extract which was chromatographed over 1.5 kg of silica

gel, 900 ml fractions being eluted as follows: frs. 1-8 (hexane), 9-12 (hexane-EtOAc, 99:1), 13-16 (hexane-EtOAc, 49:1), 17-19 (hexane-EtOAc, 97:3), 20-23 (hexane-EtOAc, 19:1), 24-30 (hexane-EtOAc, 93:7), 31-35 (hexane-EtOAc, 9:1), 36-45 (hexane-EtOAc, 17:3), 46-52 (hexane-EtOAc, 4:1), 53-60 (hexane-EtOAc, 3:1), 61-68 (hexane-EtOAc, 7:3), 69-72 (hexane-EtOAc, 13:7), 73-83 (hexane-EtOAc, 3:2), 84-92 (hexane-EtOAc, 1:1), 93-97 (hexane-EtOAc, 2:3), 98-102 (hexane-EtOAc, 1:4), 103-108 (EtOAc), 109-113 (EtOAc-MeOH) and 114-118 (MeOH).

Fr. 11(4.0 g) was rechromatographed over silica gel (100 g). Elution with hexane, EtOAc and EtOH mixtures of increasing polarity afforded 180 mg of a mixture. Further purification by TLC gave 160 mg of pure 1a. Frs. 15 and 16 were combined (5.9 g) and washed with CHCl<sub>3</sub> to give 1.7 g of a yellow mixtures of flavonoids. Pressure chromatography of a 30 mg portion of the mixed flavonoids over a polyamide column and elution with MeOH-EtOAc-HOAc (95:2.5:2.5) gave 12 mg of quercetin 3-methyl ether (2a), mp 234°C, identified by MS, UV (MeOH, MeOH-NaOMe, MeOH-AlCl<sub>3</sub>, MeOH-HCl, MeOH-NaOAc and MeOH-H<sub>2</sub>BO<sub>3</sub>) and <sup>1</sup>H-NMR spectrometry, and 5 mg of luteolin (2b) mp. 262°C, identified in the same manner. Concentration of the CHCl<sub>3</sub> washings at reduced pressure, chromatography of the residue (4.14 g) over silica gel and elution with hexane, EtOAc and EtOH mixtures of increasing polarity gave 1.2 g of solid which was washed with Et<sub>2</sub>O. On being kept in the refrigerator the ether washings deposited 53.6 mg of 1b.

Dry column chromatography (6 g of silica gel deactivated with 10% H<sub>2</sub>O) of fr. 17 (3.10 g) and elution with hexane-EtOAc (2:3) gave 2.11 g of material which on rechromatography over 42.2 g of silica gel and elution with hexane, EtOAc and MeOH mixtures of increasing polarity afforded 52.6 mg of 1d which after recrystallization from H<sub>2</sub>O-MeOH 19:1

**Table 1.** <sup>1</sup>H-NMR spectra of compounds 1a-1d (CDCl<sub>3</sub>).

H	1a*	1b**	1c**	1d**
2 $\alpha$	2.52ddd(12,4,3.5)	{ 2.3-	{ 2.3-	{ 2.3-
2 $\beta$	2.73ddd(12,12,3.5)	{ 3.0m	{ 3.0m	{ 3.0m
3 $\alpha$	2.84brddd(12,12,3.5)	{	{	{
3 $\beta$	2.21ddd(12,5,3)			
5	5.17brdq(10.5,1.5)	5.15m	5.15m	5.15m
6	4.60dd(10.5,9)	4.57dd(10.7,10)	4.59dd	4.58dd
7	3.27ddd(10,9,3.5,3)	3.15dddd (10.7,6.6,3.3,3)	3.20dddd	3.21dddd
8	5.10ddd(10,4,3)	5.1m	5.1m	5.1m
9 $\alpha$	2.03dd(16.5,3)	2.0m	2.0m	2.0m
9 $\beta$	2.31dd(16.5,4)	2.3m	2.3m	2.3m
13a	6.21d(3.5)	6.38d(3.3)	6.26d	6.27d
13b	5.67d(3)	5.80d(3)	5.78d	5.84dd
14***	1.22s	1.28s	1.26s	1.22
15***	1.91d(1.5)	1.91d(1.3)	1.90d	1.90d
3'a	6.22quint(1)	4.02d(11)	3.95d(12)	
3'b	5.69brs	3.68d(11)	3.57d(12)	
4'***	1.99t(1)	1.58s	1.48s	1.50s
Ac***		-		2.08s
OH	3.78br			

\* 500 MHz

\*\* 80 MHz

\*\*\* Intensity three protons

Table 2.  $^{13}\text{C}$ -NMR spectra of compounds 1a-1d ( $\text{CDCl}_3$ )

C	1a*	1b*	1c**	1d**
1	215.52s	214.70s	214.27s	214.73
2	34.18t <sup>a</sup>	34.23t <sup>a</sup>	34.20t <sup>a</sup>	34.20t <sup>a</sup>
3	36.93t <sup>a</sup>	35.83t <sup>a</sup>	35.89t <sup>a</sup>	35.76t <sup>a</sup>
4	142.16s	142.06s	142.35s	142.22s
5	124.82d	124.72d	124.53d	124.66d
6	76.64d	76.12d	76.05d	76.15d
7	47.59d	47.34d	47.34d	47.27d
8	72.38d	74.69d	74.98d	74.52d
9	35.26t <sup>a</sup>	35.77t <sup>a</sup>	35.85t <sup>a</sup>	35.76t <sup>a</sup>
10	78.57s	78.73s	78.81s	75.95s
11	136.04s <sup>b</sup>	135.95s	136.17s	135.97s
12	169.69s	169.30s	169.27s	169.47s
13	123.29t	123.89t	123.88t	123.94t
14	29.81q	29.31q	29.45q	29.71q
15	18.05q	17.51q	17.62q	17.75q
1'	166.26s <sup>b</sup>	172.29s	174.99s	172.78s
2'	135.94s	74.78s	77.45s	75.95s
3'	126.68t	50.68t	68.08t	68.80t
4'	18.05q	22.97q	21.78q	22.30q
1''				170.25s
2''				20.80q

\* at 20 MHz

\*\* At 55 MHz

<sup>a,b</sup> Assignments in same column with same superscript may be interchanged.

melted at (152-153) °C. Recrystallization from water of a 300 mg portion of fr. 18 (2.5 g) gave 60 mg of 1c, mp. (160-162) °C. (6R\*, 7R\*, 8S\*, 10S\*)-10-Hydroxy-8-methacryloxy-1-oxogermacra-4,11(13)-dien-6,12-olide (1a). Mp (180-182) °C; MS m/z (rel.int.) 349 ( $\text{C}_{19}\text{H}_{24}\text{O}_6^+ + \text{H}$  by intramolecular hydrogen transfer, 30), 331 (6), 262 (100), 244 (47); IR  $\nu$   $\text{cm}^{-1}$  3440, 1760, 1712, 1698, 1670-1635;  $^1\text{H}$ -NMR spectrum in Table 1;  $^{13}\text{C}$ -NMR spectrum in Table 2.

(6R\*, 7R\*, 8S\*, 10S\*)-10-Hydroxy-8-(2-methyl-2-hydroxy-3-chloropropanoyloxy)-1-oxo-germacra-4,11(13)-dien-6,12-olide (1b). Mp (163-164) °C; MS m/z (rel.int.) 403 (11) and 401 (35,  $\text{C}_{19}\text{H}_{25}\text{O}_7\text{Cl}$ ), 263 (100), 245 (66), 227 (23), 205 (28); IR  $\nu$   $\text{cm}^{-1}$  3610, 3410, 1760, 1740, 1705, 1665-1625;  $^1\text{H}$ -NMR spectrum in Table 1;  $^{13}\text{C}$ -NMR spectrum in Table 2.

(6R\*, 7R\*, 8S\*, 10S\*)-10-Hydroxy-8-(2-methyl-2,3-dihydroxy-propanoyloxy)-1-oxogermacra-4,11(13)-dien-6,12-olide (1c). Mp (160-162) °C (from  $\text{H}_2\text{O}$ ); MS m/z (rel.int.) 383 ( $\text{C}_{19}\text{H}_{26}\text{O}_8^+ + \text{H}$ , by intramolecular hydrogen transfer, 30), 365 (8), 281 (28), 263 (100), 244 (58); IR  $\nu$   $\text{cm}^{-1}$  3420, 1763, 1722, 1702, 1670-1650;  $^1\text{H}$ -NMR spectrum in Table 1;  $^{13}\text{C}$ -NMR spectrum in Table 2.

(6R\*, 7R\*, 8S\*, 10S\*)-10-Hydroxy-8-(2-methyl-2-hydroxy-3-acetoxy-propanoyloxy)-1-oxogermacra-4,11(13)-dien-6,12-olide (1d). Mp (152-153) °C (from  $\text{H}_2\text{O}$ , MeOH

Table 3. NOE difference spectrum of 1a.

Irradiated	Observed(%)
H-2 $\beta$	H-2 $\alpha$ (9.1) H-14 (2.7)
H-3 $\alpha$	H-3 $\beta$ (29.3), H-5 (13.1)
H-5	H-3 $\alpha$ (5.8), H-7 (14.7)
H-6	H-8 (9.8), H-15 (10.7)
H-7	H-5 (13.3)
H-8	H-6 (6.6), H-9 $\alpha$ (1.8), H-9 $\beta$ (2.2)
H-9 $\alpha$	H-8 (3.2), H-9 $\beta$ (12.1)
H-14	H-2 $\beta$ (1.3), H-9 $\beta$ (2.3)

19:1); MS m/z (rel.int.) 425 ( $\text{C}_{21}\text{H}_{28}\text{O}_9$ ,  $\text{M}^+ + \text{H}$  by intramolecular hydrogen transfer, 100), 407 (15), 263 (75), 245 (36); IR  $\nu$   $\text{cm}^{-1}$  3450, 1755, 1735, 1730, 1710, 1670-1650;  $^1\text{H}$ -NMR spectrum in Table 1;  $^{13}\text{C}$ -NMR spectrum in Table 2.

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